

Mn₁₂ single molecule magnets deposited on μ -SQUID sensors: the role of interphases and structural modifications†

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Direct measurements of the linear ac susceptibility and magnetic relaxation of a few Mn₁₂ monolayers deposited on a μ -SQUID sensor are reported. In order to integrate the molecules into the device, DPN has been the technique of choice. It enabled the structuration of the molecules on the most sensitive areas of the sensor without the need for any previous functionalization of the molecule or the substrate, while controlling the number of molecular units deposited on each array. The measurements reveal that their characteristic SMM behaviour is lost, a fact that is attributed to molecular distortions originated by the strong surface tensions arising at the molecular interphases.

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Introduction

Since their discovery in the early 90s,¹ single-molecule magnets (SMMs) have aroused interest as an alternative and complementary approach to replace inorganic solids on magnetic devices and applications.² These molecular clusters show slow magnetic relaxation and magnetization hysteresis³ thanks to the combination of high-spin ground states and strong axial magnetic anisotropy.⁴ In addition, they also exhibit other interesting phenomena such as quantum tunnelling of the magnetization.⁵ Advantages of SMMs over their inorganic counterparts are their small size, below a few nanometres, their close to perfect monodispersity, and the fact that they can be obtained in large quantities in a reproducible manner at reasonably low costs and manipulated in a large variety of solvents and matrices. Yet, applications remain elusive. A fundamental issue that still needs to be addressed is the development of strategies for the evolution of bulk materials to nanostructures organized over the surfaces of solid substrates while retaining their characteristic magnetic behaviour.

The archetypical case for all these years has been the dodecamanganese acetate cluster (Mn₁₂-ac) and its derivatives.⁶ The number of successful techniques employed for their structuration on surfaces (with a few exceptions, under vacuum conditions) is considerably large.⁷ However, preservation of their SMM behaviour still remains a controversial and challenging issue. Other complexes, such as the Fe₄ derivatives, provide promising alternatives to circumvent this difficulty.⁸ Though, the large number of studies aimed to integrate Mn₁₂ on surfaces represents an excellent scenario to discuss some of the main challenges still lying ahead. Such information is crucial for an efficient design of future hybrid devices based on SMMs.

For instance, direct charge transfer to the substrate has been reported as one of the main mechanisms leading to the loss of SMM behaviour, upon chemical modification of the magnetic core. O'Shea *et al.* have observed a significant charge transfer between gold and a monolayer of Mn₁₂-ac molecules even after protection of the magnetic core with large benzoate ligands.⁹ Sessoli *et al.* also observed a partial reduction to Mn(II) on Au(111) for two different Mn₁₂ thio-derivatives.¹⁰ More successful were Pennino *et al.*¹¹ These authors organized monolayers of Mn₁₂ with longer thiol-terminated ligands on Au(111), and found that a large fraction of molecular cores retain their structural integrity. Such divergence of results between different experiments can be attributed to the role that the isolating organic layer plays in decoupling the core magnetic ions from the surface, as recently shown by Forment-Aliaga *et al.*¹² and Voss *et al.*¹³ One of the best alternatives to overcome these chemical restrictions is to use substrates with different charge transfer properties. For this, Ternes *et al.* recently demonstrated the controlled deposition of fragile Mn₁₂

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SMMs on thin-insulating surfaces such as boron nitride (BN) and the feasibility for preserving their quantum magnetism.¹⁴ Some studies performed with different substrates, such as HOPG,¹⁵ glass¹⁶ or silicon,¹⁷ have also been reported.

Further studies have shown that preserving the core is a necessary but not sufficient condition to retain the SMM relaxation behaviour of Mn₁₂ clusters. Local symmetry changes at the Mn sites, arising from the interplay of attractive and repulsive molecule–surface interactions, can also take place. This has been directly inferred from the modification of the spectral line shape of X-ray absorption spectra (XAS)¹⁸ or by atomic force microscopy (AFM) force-volume imaging experiments.¹⁹ Such symmetry changes prompted modification of local Jahn–Teller distortions in a rather uncontrolled manner and, therefore, modification of their magnetic relaxation through variations of the magnetic anisotropy constant *D*.

In any case, testing the magnetic properties of the Mn₁₂ molecules on the surface clearly represents the best alternative approach to assess their behaviour. So far, most of the techniques applied to this end have been based on indirect probes, such as Scanning Tunneling Microscopy (STM), β -detected Nuclear Magnetic Resonance (β -NMR) or mainly X-Ray Magnetic Circular Dichroism (XMCD).^{10,11,17–20} However, direct measurement of the magnetic properties with a μ -SQUID sensor has remained so far elusive due to the insufficient sensitivity and the lack of experimental techniques for their proper integration. This represents a key challenge in the field nowadays.

In this work, the ac magnetic susceptibility of Mn₁₂ benzoate (Mn₁₂bz) is directly measured by depositing a few layers of the molecular material on the most sensitive areas of a miniaturized Superconducting Quantum Interference Device (μ -SQUID) susceptometer. The frequency-dependent blocking of the ac linear response is one of the characteristic traits of a SMM. A variation of the frequency and temperature over wide ranges enables the exploration of spin relaxation phenomena and provides information on the magnetic energy level schemes. Therefore, the implementation of these devices and their application to the study of molecular nanostructures provides a relatively simple and powerful method for directly characterizing SMMs deposited on a substrate. Integration of the molecules onto the optimum region of the device has been addressed by means of Dip-Pen Nanolithography (DPN). It has already been shown that DPN is an excellent technique to accurately deposit magnetic nanoparticles on a μ -SQUID without the need for any previous functionalization,²¹ while controlling the number of molecular units deposited on each array.²²

Experimental

Materials

Mn₁₂bz was synthesized by following synthetic protocols published previously.²³ Glycerol (GC, $\geq 99\%$) and *N,N*-dimethylformamide (DMF, anhydrous, 99.8%) were purchased from Sigma-Aldrich, whereas ethanol absolute (Sps, $>99.8\%$) and acetonitrile (190 Sps, far UV/gradient quality, $>99.9\%$) were purchased from Romil. All these materials were used as received without further purification.

Substrates

Silicon bearing native oxide layer (Si/SiO₂) substrates were prepared by initially cutting silicon wafers into $0.5 \times 0.5 \text{ cm}^2$ pieces. Then, each Si/SiO₂ substrate was washed in an ultrasonic bath for 10 min with acetonitrile, ethanol, and Milli-Q water and dried by blowing nitrogen gas. Carbon-coated TEM grids, carbon type-B, 200 mesh copper grids were purchased from Ted Pella, Inc.

Dip-pen nanolithography experiments

DPN experiments were performed with an Nscriptor™ DPN System (from NanoInk, Inc.). Environmental conditions of $\sim 40\%$ relative humidity and room temperature were kept constant during all patterning processes with an integrated environmental chamber as part of the Nscriptor DPN System. Tips specially designed for writing purposes were used in all the DPN experiments. Specifically, commercial silicon nitride Type M Probe Arrays and Type A Single Pens (from NanoInk, Inc.) with a spring constant of 0.5 N m^{-1} and 0.1 N m^{-1} , respectively, were used. To coat the tip, a microfluidic ink delivery chip-based system (Inkwell, from NanoInk, Inc.) was used. The inkwells contain several reservoirs that are filled with the desired ink solution with the help of a micropipette. Afterwards the ink is transferred to the microwells where the tips are dipped for a given time. The excess of ink coating the tip is reduced by bringing the freshly coated tip in contact with a Si/SiO₂ substrate and creating spots with an average diameter of several micrometers, which quickly decreased in size. After having deposited a few micrometer-sized spots, it is possible to start writing uniform dots in a very controllable manner. To create patterns over large areas of the Si/SiO₂ surface, MP tips were chosen, since once coated they allow creating 12 identical structures in one single step on Si/SiO₂. Tips were repeatedly dipped into the reservoirs and used for writing until depletion of the ink, allowing patterning of large arrays over a span of hours.

Characterization methods

Dynamic noncontact mode AFM images were recorded on an Agilent 5500 AFM/SPM microscope. For this purpose, noncontact high-resonance frequency with reflex coating (NCHR) tips were used, which are PPP-NCH silicon point probes with a spring constant of $\sim 42 \text{ N m}^{-1}$ and a resonant frequency of $\sim 330 \text{ kHz}$ (from Nanosensors). AFM image processing and rendering was done with Gwyddion data analysis software. HR-TEM images, EDX analysis and SAED were carried out on a JEM-2100 microscope (from JEOL LTD) with an acceleration voltage of 80 to 200 kV. Additional TEM images were taken with a JEM-1400 microscope (from JEOL LTD) with an acceleration voltage of 40 to 120 kV. Static contact angle measurements, based on the sessile drop method, were performed using a drop shape analyzer Easy Drop combined with Drop Shape Analysis DSA1 (from Kruss). In the experiment a $3 \mu\text{l}$ droplet of the ink solution was deposited on the target substrate surface using a micropipette. The contact angle was measured on five different regions of the substrate and an average value was obtained from all the



measures. SEM images were recorded on a FEI dual beam Helios 600. Grazing incidence XRD measurements were performed on a Panalytical X'Pert Pro X-ray diffractometer by using constant grazing incidence (0.5°) and parallel beam optics. The analysed area was about $5 \times 20 \text{ mm}^2$. IR spectra were recorded on a TENSORTM 27 FT-IR spectrometer with an ATR accessory from Bruker Optics. XPS experiments were performed using a PHI 5500 Multitechnique System (from Physical Electronics) with a monochromatic X-ray source (Aluminium K α line of 1486.6 eV energy and 350 W), placed perpendicular to the analyzer axis and calibrated using the 3d $_{5/2}$ line of Ag with a full width at half maximum (FWHM) of 0.8 eV. The area under analysis was a circle of 0.8 mm diameter. The energy resolutions were 187.5 eV of Pass Energy and 0.8 eV per step for the general spectra and 23.5 eV of Pass Energy and 0.1 eV per step for the spectra of the different elements. All XPS measurements were made in an ultra high vacuum (UHV) chamber with pressures between 5×10^{-9} and 2×10^{-8} torr.

MicroSQUID susceptometer

A new generation of integrated μ -SQUID susceptometers was developed at PTB.²⁴ The devices are made of Nb deposited on Si/SiO $_2$ wafers and they are fabricated by optical lithography. In order to enhance the coupling between the sample and the sensor with respect to that achieved with previous designs,^{25,26} the diameters of the two circular loops forming the SQUID were reduced to 30 μm . The ac excitation coil allows the application of ac magnetic fields with amplitudes of up to 10 Oe to be compared with amplitudes of approximately 0.1 Oe available with our first and second generation μ -SQUIDs.^{25,26} In addition, the device was designed with close to perfect reflection symmetry, in order to minimize the output signal when no sample is present. The instrument exhibits a white flux noise of $1.3 \mu\Phi_0 \text{ Hz}^{-1/2}$.

Results

Surface structuration

In a typical DPN experiment, an Atomic Force Microscope (AFM) tip is first functionalized with a solution that contains the molecular entities (referred to as ink). Then, the coated tip is used to directly transfer the ink, in the form of droplets, to the surface. Notwithstanding, the ink solvent plays an important role in the integration process and should fulfil the following characteristics: (i) an adequate viscosity to ensure homogeneous coating of the tip and a controlled transfer of the material; (ii) a high boiling point to maintain the liquid state over the whole writing process (iii) a proper solvent–surface wettability balance to guarantee both the controlled delivery of the ink and the maintenance of the structure once the ink is dried and (iv) a high solubility and stability of Mn $_{12}$ bz.

In view of these considerations, our first experiments were devoted to finding a proper solvent for the structuration of Mn $_{12}$ bz on Si/SiO $_2$ substrates. This is the material used in the sensitive μ -SQUID areas. Most of the above described requirements were met by a binary mixture of dimethylformamide

(DMF) : glycerol. DMF exhibits a high boiling point (b.p. 153°C) and a relatively high viscosity, both enhanced by adding glycerol to the ink solution (b.p. 290°C). The glycerol ratio was tuned by measuring for different solutions both the contact angle (CA) of the deposited drops, which depends on the electrical polarity of the mixture, and the ability to transfer the solution from the AFM tip to the substrate (viscosity).

CA of μl droplets deposited on Si/SiO $_2$ substrates significantly increases with the glycerol concentration; values of $\sim 10^\circ$, $\sim 18^\circ$, $\sim 20^\circ$ and $\sim 24^\circ$ were measured for, respectively, pure DMF, and mixtures of DMF + 5% v/v glycerol, DMF + 15% v/v glycerol and DMF + 20% v/v glycerol (see Fig. S1 of the ESI†). A larger CA upon glycerol concentration increase allows for a better control on the resulting morphologies, because it helps maintaining the pattern shape intact during the drying of droplets. However, an increase in the glycerol concentration also induces a considerable increase of the viscosity that disrupts the ink transfer from the AFM tip. Therefore, the optimum DMF : glycerol ratio found, after different deposition experiments, is a compromise between 5 and 10% v/v. The stability of the molecular system in such a binary mixture was confirmed by spectroscopic experiments, more specifically by attenuated total reflectance-Fourier transform infrared spectroscopy (ATR-FTIR) and X-ray photoelectron spectroscopy (XPS) (see ESI, S2 and S3,† respectively).

A dark brown-colored solution of Mn $_{12}$ bz ($10\text{--}20 \text{ mg mL}^{-1}$) in a DMF : glycerol (95 : 5) mixture was freshly prepared and used as ink for coating the tip (for more details see the Experimental section). The coated tip was brought into contact with the surface to deposit drops of controlled size. Mn $_{12}$ arrays can then be fabricated by traversing the tip over the desired area. Reproducible and uniform in size structures were obtained in this way. As a representative example, Fig. 1a shows a well-defined array of Mn $_{12}$ bz made of 20×20 dots spaced by 2 μm . The height profile analysis of the dots reveals a high uniformity in size, with an average diameter of $555 \pm 39 \text{ nm}$ and a height of $6 \pm 1 \text{ nm}$. Considering the dimensions expected for the disk-like Mn $_{12}$ bz molecule, of approximately $2.1 \times 1.2 \text{ nm}$ (obtained by molecular modelling), and taking into account the presence of some residues remaining after solvent evaporation, the height of these dots is in agreement with the distribution of the molecules forming no more than 2–4 molecular layers. To illustrate the effect of the solvent on the final dimensions and structure of the arrays, Fig. 1b shows results obtained, under the same conditions, with a Mn $_{12}$ bz-free mixture of DMF and 5% v/v of glycerol. AFM analysis of the structures indicates a surface topography of up to 1 nm, attributed to the presence of some solvent residues that can be clearly differentiated from our material.

Further control over parameters such as the ink loading of the tip, its movement over the surface and the tip-substrate contact time allowed us to gain control over the deposited structural motifs. As a representative example, Fig. 1c shows a pattern pre-designed to fit inside the internal diameter of the μ -SQUID (27 μm , see Fig. 1d). This pattern consists of concentric circles of dots separated by 1 μm .

Grazing incidence X-ray diffraction (GI-XRD) was used to determine the degree of crystallinity of the deposits.



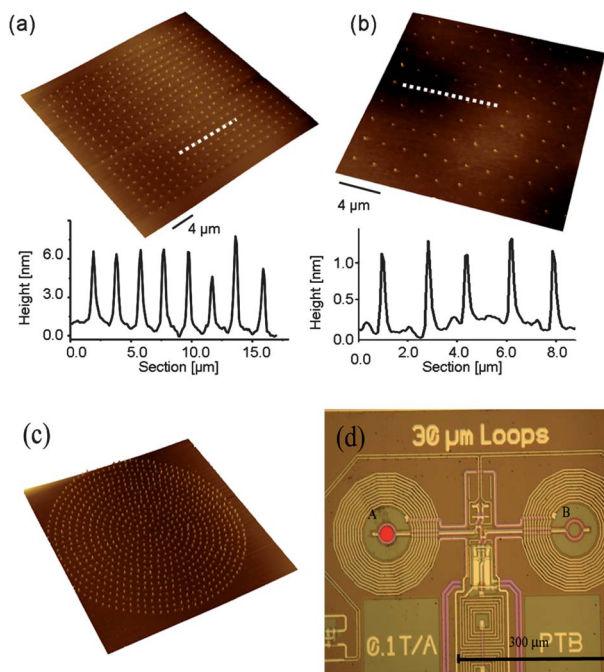


Fig. 1 Top: arrays fabricated by DPN on Si/SiO₂ substrates. 3-D AFM topography image and height profiles recorded along the white dashed lines in the corresponding AFM images are shown. The specifications for each particular case are: (a) ink composition: Mn₁₂bz (10 mg mL⁻¹) in DMF and 5% v/v of glycerol; (b) ink composition: DMF and 5% v/v of glycerol. Bottom: (c) 3-D AFM topography image of the circular pattern specially designed to fit inside the pickup coil of the susceptometer. The external diameter is 25 μm and the distance between dots is 1 μm. The array was fabricated on Si/SiO₂ using the same experimental conditions as in (a). (d) Optical image of the two pickup coils that correspond to the sensitive areas of the microSQUID sensor. The sample to be measured can be located on either the left hand side (A), wherein as an example a sample is indicated with a red circle, or the right hand side (B).

Experiments were performed on two different samples: a thin-film and a structured sample. The thin-film sample was prepared by drop casting Mn₁₂bz ink on a substrate, while the structured sample was fabricated by DPN. For the latter, the size of the droplets was adjusted to ~1 μm and the deposition process was repeated over extended areas in order to cover up to a few square millimetres, which are required to have enough sensitivity with this experimental technique (see ESI, S4–S5†).

The absence of any clear diffraction peak at low 2θ angles indicates that both samples are amorphous or that they possess a very small degree of crystallinity (see Experimental section). It could also be possible, especially in the case of the structured samples (arrays), that the signal is too small to be detected by the present XRD setup. Further evidence for the amorphous nature of Mn₁₂bz deposits was obtained by high-resolution transmission electron microscopy (HR-TEM) experiments. It should be noted that HR-TEM images were taken with a minimum electron dose, in order to avoid damaging the sample with the beam. Typical HR-TEM images of the arrays are shown in Fig. 2. Even though several dots were inspected over a range of samples, no lattice fringes were observed in any of them (Fig. 2c, inset). These results suggest that Mn₁₂bz droplets deposited on surfaces and exposed to ambient conditions tend

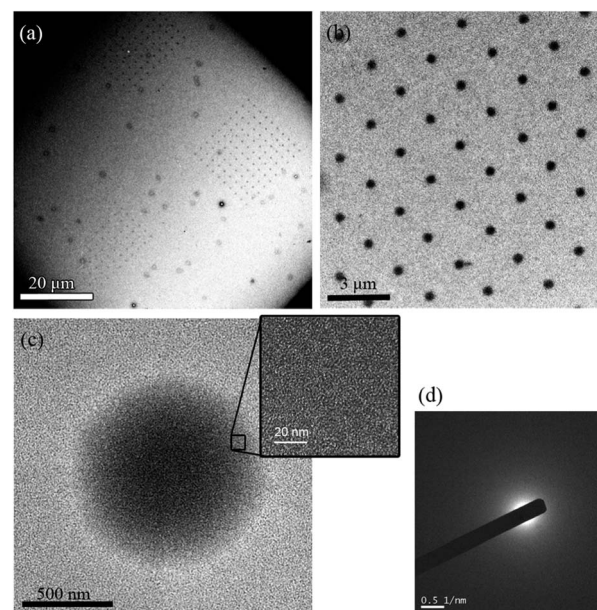


Fig. 2 HR-TEM images of (a) three different arrays of Mn₁₂bz fabricated on a carbon-coated TEM grid and (b) magnification view of one of the arrays. Ink composition: Mn₁₂bz (10 mg mL⁻¹) in DMF and 5% v/v of glycerol. (c) Details of one of the dots of the array. (The inset shows the magnification view of a region inside the dot). (d) Electron diffraction pattern obtained from a region inside the dot.

to dry as an amorphous material. This is further supported by the presence of a diffuse halo in the selected area electron diffraction (SAED) patterns obtained from regions inside all dots under study (see Fig. 2d).

Integration of molecular samples into μ -SQUID sensors

The μ -SQUID susceptometer used for these studies²⁴ consists of the μ -SQUID sensor itself and an ac excitation coil, which generates a close to uniform magnetic field $0.01 \text{ Oe} < h_{ac} < 7 \text{ Oe}$ oscillating with time at frequency ω . As Fig. 1d shows, the μ -SQUID has a gradiometric design, with two circular loops wound in opposite directions and linked together by two Josephson junctions. The sensor converts the net magnetic flux through these two coils, which arises from any imbalance between the flux “seen” by each of them, into an output voltage. An ac output voltage can therefore be induced by placing a magnetic sample into one of these coils, and applying an ac magnetic field h_{ac} . The areas delimited by the two SQUID coils represent the sensitive, or active, areas of the susceptometer. A small, but unavoidable, imbalance between the two coils gives also rise to a finite output signal for the empty susceptometer. This contribution has been measured independently and can be subtracted from the total output signal, in order to obtain the susceptibility of the sample. Yet, this effect limits the minimum detectable voltage to approximately 10^{-6} V , as it sets a minimum scale for the lock-in amplifier that is used to read-out the SQUID. Under these conditions, it is crucial to optimize the flux coupling of the sample with the SQUID, as this coupling determines the output signal generated by a given sample. This goal is achieved by uniformly depositing the sample inside the coil. For the measurements on Mn₁₂bz samples, we made use of



two devices with the same design but with slightly different background signals. We shall refer to them in what follows as **SQUID1** and **SQUID2**.

The two Nb coils forming the SQUID are circular-shaped with external and internal diameters of 35 μm and 27 μm , respectively (see Fig. 3). The coil wire has a cross-section of approximately 5 μm \times 450 nm (a more detailed description of the sensor is provided in the ESI†). Five different Mn_{12}bz samples were integrated into one of the μ -SQUID coils. Three of these samples were integrated by DPN, as described below, using the experimental conditions already optimized for the controlled deposition of Mn_{12}bz on bare Si/SiO_2 (a mixture of DMF and 5% v/v of glycerol). In addition, two “bulk-like” samples were also measured, for comparison purposes, under the same conditions. The first of these consisted of a micron-sized crystal of Mn_{12}bz placed directly onto the coil (**B1**). The last sample (**B2**) was microcrystalline powder obtained after evaporation of a DMF solution of the Mn_{12}bz . These two samples were embedded in apiezon-N grease and placed onto the μ -SQUID with the use of a home-made micromanipulator.

The first DPN sample (hereafter referred to as **DPN1**) consisted of a circular pattern of dots deposited inside the μ -SQUID coil of the **SQUID1** sensor (see Fig. 3b and c) after completely removing the excess ink at the tip. In order to do so, several spots were fabricated by the freshly coated tip on an auxiliary Si/SiO_2 surface until writing uniform dots in a very controllable manner (see Experimental section).

Afterwards, the tip was placed on top of the coil wire to create a circular pattern while keeping the tip-substrate contact time at 0.1 s. The process was repeated 2.5 times to deposit 1320 identical dots inside the coil. After the deposition, the sensor was characterized by scanning electron microscopy (SEM). The images show the effective and regular deposition of the

molecular material inside the circular sensing area of the susceptometer (see Fig. 3b). The topography of the array was investigated by performing AFM on a replica fabricated on a bare Si/SiO_2 surface (Fig. 3c).

The purpose of this method is to avoid damaging the sensor in the course of the AFM experiments. The analysis of the height profiles, shown in Fig. 3d, reveals the formation of highly uniform in size dots, with an average diameter of 249 ± 9 nm and a height of 8 ± 1 nm. These data suggest that each dot consists of approximately 3–5 Mn_{12}bz layers. Taking into account the size and number of the dots deposited, it can be estimated that about 9.0×10^7 Mn_{12}bz molecules were integrated inside the sensing area of the susceptometer (for more details on the determination of the number of molecules deposited see the ESI†).

In the subsequent deposition of sample **DPN2** on sensor **SQUID2**, the amount of Mn_{12}bz transferred to the device was increased by controlling the amount of ink removed from the tip in the previous step. For this, the freshly coated tip was brought into contact with the auxiliary Si/SiO_2 surface for a shorter time than in the previous experiment. Afterwards, the tip was again positioned on the wire of one of the pickup coils by optical control and traversed over the surface, as before, forming a circular pattern with a tip-substrate contact time of 0.1 s.

In this case, the larger volume of the droplets makes them merge together on the substrate, forming a continuous thin film that covers the entire coil (sample **DPN2**). An optical image of the pickup coil captured during the deposition process is shown in Fig. 4a. The height profile of a replica fabricated under

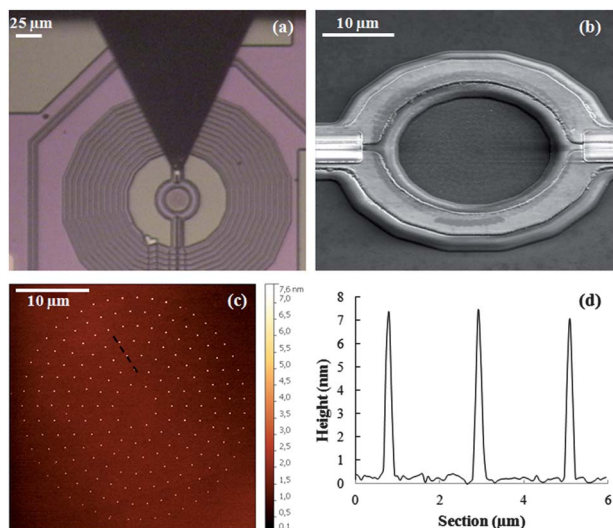


Fig. 3 (a) Optical image of the μ -SQUID susceptometer right after the deposition process (b) FE-SEM images taken at 45° tilt angle on one of the susceptometer's pick-up coils after integration of sample **DPN1**. (c) AFM topography image of a similar array fabricated on Si/SiO_2 using the same experimental conditions used for **DPN1**. (d) Height profile measured by scanning the tip along the black dashed line in image (c).

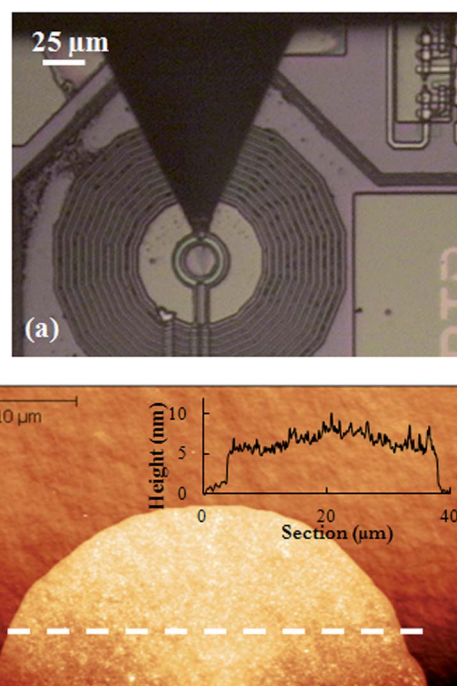


Fig. 4 (a) Optical image of the μ -SQUID susceptometer taken after depositing sample **DPN2**. (b) AFM topography image of a deposit formed on Si/SiO_2 under identical conditions to those used for sample **DPN2** and (inset) cross-section recorded along the white dashed line in image (b).



identical conditions but on a bare Si/SiO₂ surface indicates a height of ~ 8 nm (Fig. 4b). This profile is in agreement with that expected from a Mn₁₂bz multilayer formed by 3–5 molecular layers. The number of molecules deposited on the sensing area is estimated to be of the order of 7.2×10^8 (see ESI†).

Finally, when the same circular pattern is repeated without previously removing any excess ink from the tip, a large droplet completely covering the coil area with the Mn₁₂bz ink is obtained (sample DPN3). For this, the tip was positioned on the centre of the coil of SQUID1 by optical control and brought into contact with the surface to deliver a sufficient amount of ink in a single step. An image of this “bulky” sample as it looked right after the DPN deposition process is shown in Fig. 5. The topography of the dry droplet was measured by AFM on a similar sample fabricated under identical conditions but on a bare Si/SiO₂ surface. The topographic profile shows an average height of ~ 15 – 20 nm, in agreement with the distribution of the molecules into up to 6–8 molecular layers.

Magnetic susceptibility measurements

Fig. 6 shows the output signals recorded, at $T = 4.2$ K and with the μ -SQUID susceptometer SQUID1, from samples B1-2, DPN1 and DPN3. Data measured for sample DPN2 using SQUID2 are given as part of the ESI (Fig. S7†). The response of the crystalline

sample (B1) shows the existence of two relaxation processes. The frequency-dependent contribution observed at low frequencies (~ 50 Hz) is associated with relaxation of standard (or “slow”) Mn₁₂bz clusters while the second one (~ 300 kHz) comes from “fast” relaxing clusters, which represent the majority in this compound.²³ The very different relaxation rates observed at $T = 4.2$ K are in agreement with the different activation energies involved in the spin reversal of these isomers. Similar results were obtained for sample B2 (Fig. 6A and B); the two magnetic relaxation processes remain, that is, Mn₁₂bz clusters largely preserve their magnetic memory. However, the imaginary susceptibility peaks of sample B2, thus also the steps shown by the in phase component, are considerably broader than those of the single crystal, thus suggesting the presence of larger dispersions of relaxation times for both isomers. This is understandable, as the fast recrystallization experienced by sample B2 is bound to increase the density of lattice defects and therefore leads to distribution of molecular environments.

The responses of two samples deposited by DPN (DPN-1 and DPN-3) are shown in Fig. 6C and D. The dependence of V' and V'' on the frequency differs strikingly from that found for the precursor crystalline material B1. In particular, V' is close to constant above the background signal of the empty susceptometer, while the imaginary component V'' vanishes. The same behavior is observed for sample DPN2 (see Fig. S7 of the ESI†).

The upturns observed above 10^4 Hz can be fully ascribed to the background signal of the μ -SQUID susceptometer. These results show that Mn₁₂bz molecular nanomagnets deposited on silicon substrates remain in equilibrium up to very high

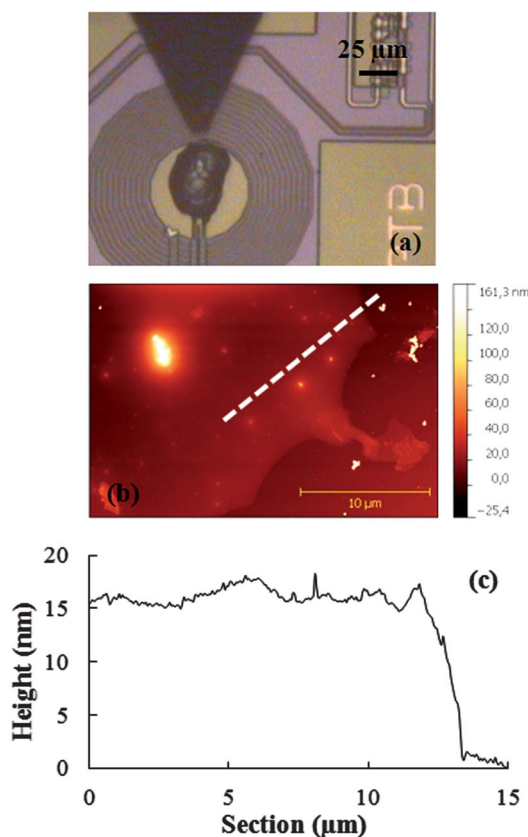


Fig. 5 Deposition of a continuous Mn₁₂bz thin-film covering the whole coil area (sample DPN3). (a) Optical image taken during the integration process. (b) AFM topography image of the replica sample fabricated, under the same conditions as in (a), onto a broken μ SQUID susceptometer. (d) Height profile measured along the white dashed line in (b).

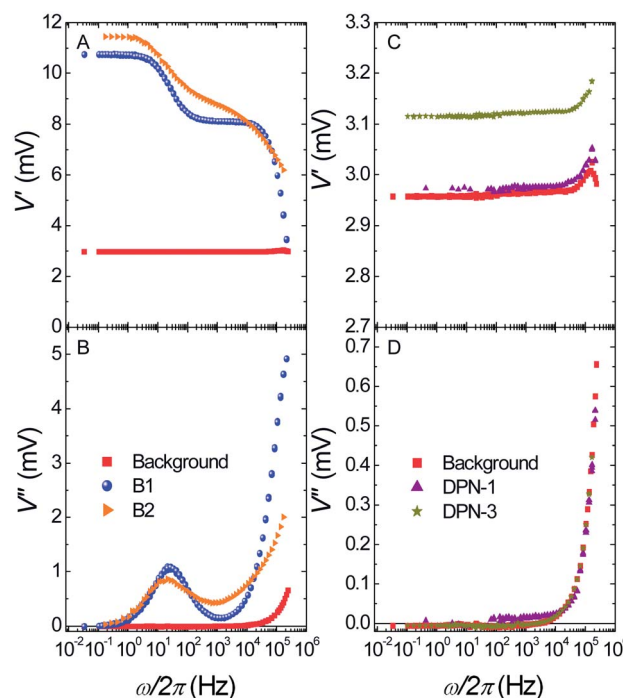


Fig. 6 Output signals measured on empty μ -SQUID susceptometer SQUID1 (background) and on the same device after the deposition of several Mn₁₂bz samples. Top: in-phase components. Bottom: out-of-phase components.



frequencies, that is, their magnetic relaxation times become much faster than those in the bulk. It is worth noting that the enhanced magnetic relaxation is not associated with conversion of “slow” into “fast” ones. Instead, the susceptibility data show that drastic changes in the magnetic relaxation mechanisms occur for both isomers.

Before trying to discuss their physical significance, it is important to first make sure that these signals come indeed from the molecular samples. In order to check this question, we have computed, using the commercial simulation software COMSOL, the magnetic flux that an ensemble of Mn_{12}bz molecules couples with one of the μ -SQUID pick-up coils. The model used for these calculations is schematically depicted in Fig. 7A that shows also the spatial distribution of the magnetic field derived from these calculations. The sample is a homogeneous disk, with a density equal to that of Mn_{12}bz but with magnetic easy axes oriented at random, located inside one of the pick-up coils and subjected to the same magnetic field $h_{\text{ac}} = 25$ mOe as that used in the experiments. The number of molecules (N) is varied by changing the height h of the disk. For low N ($<10^9$), simulations performed on different sample geometries, *e.g.* on samples formed by discrete dots, give results that are in 30% agreement with those obtained for a continuous disk.

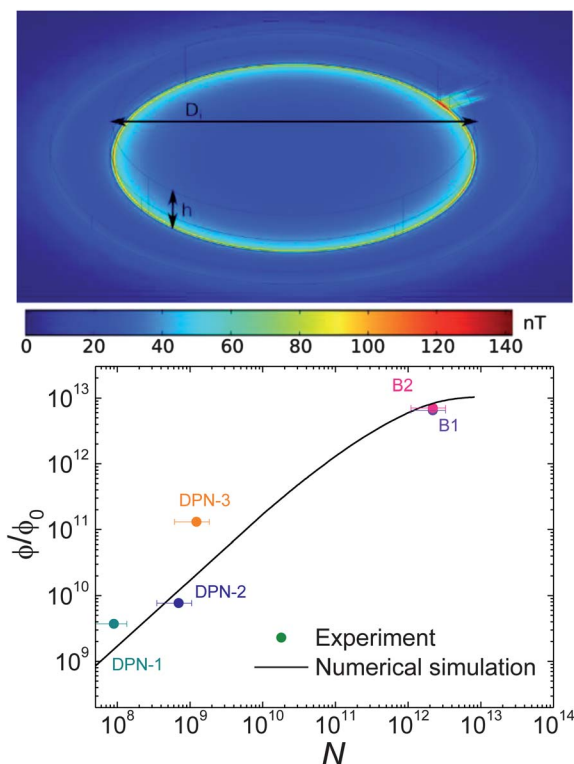


Fig. 7 Top: theoretical color plot of the magnetic field intensity generated by a homogeneously magnetized cylinder (height h , diameter $D_1 = 26 \mu\text{m}$) located inside one of the μ -SQUID pick-up coils (thin black lines). These numerical calculations are used to determine the magnetic flux ϕ coupled with the device. Bottom: dependence of ϕ as a function of the number of Mn_{12}bz molecules deposited on one of the pick-up coils. The solid line is calculated using the theoretical model described above and varying h between 8 nm ($N = 10^9$) and 80 nm ($N = 10^{13}$). The prediction for $N < 10^9$ is a linear extrapolation of these results.

Therefore, we conclude that approximating the sample with a continuous disk must be valid for the range of N values that we consider here.

The dependence of the equilibrium magnetic signal on N is shown in Fig. 7. The data are taken from the low- ω limits of frequency-dependent measurements, whereas the solid line represents the results of the above model calculations. In these calculations, the equilibrium susceptibility of each molecule is calculated as $\chi_T = g^2 \mu_B^2 S^2 / 3k_B T$, with $g = 2$ and $S = 10$. Considering the experimental uncertainties, calculations reproduce very well the measured values. Then, it can be safely concluded that the output signals we measured correspond to the magnetic response of the molecules deposited by DPN. The fact that the equilibrium susceptibility does not change much with respect to that expected for Mn_{12}bz suggests also that the molecular magnetic moments (thus g and S) are preserved.

Discussion

Suitable experimental conditions for the DPN deposition of Mn_{12}bz molecules on Si/SiO₂ substrates have been established. Surface characterization techniques show that the molecules retain their structural identity and that there is no significant charge transfer from the surface to the molecules, as previously observed for other Mn_{12} derivatives deposited on SiO₂.²⁷ This has enabled us to integrate such SMMs on the sensitive areas of a μ -SQUID sensor and to directly measure their magnetic relaxation. These measurements reveal that the magnetization relaxation differs from that observed for the bulk material. Similar results were previously observed for a monolayer of Mn_{12} on silicon (Si) studied using a complementary technique such as β -NMR.²⁸ Since charge transfer effects and/or chemical degradation upon surface deposition can be discarded, environment effects must be considered for a cautious analysis of the drastic changes in spin dynamics. Among them are (I) crystal lattice modifications, *i.e.* modifications in the acoustic phonon spectrum and/or (II) confinement effects upon nano-structuration. Both possibilities are analyzed in more detail in what follows.

(I) Crystal lattice modifications

Modification of spin-phonon relaxation mechanisms may arise from changes in the medium stiffness and density with respect to the bulk sample induced by the amorphous character of the DPN deposits and/or possible solvate molecule remnants.²⁹ However, previously reported Mn_{12} bulk studies discard this possibility. Indeed, studies performed under two extreme conditions (a crystalline sample and a nearly amorphous material) have shown that long-range crystalline disorder affects little the anisotropy, magnetic relaxation, and magnetic quantum tunneling of these materials.³⁰ In fact, amorphous Mn_{12} -based sub-50 nm spherical particles have been shown to retain the SMM behaviour with an activation energy $U = 63(1)$ K, which is close to the value of $U = 65(1)$ K found in bulk crystals. In both studies, only a subtle influence was detected; the faster tunneling rates measured for the amorphous material can be



accounted for by changes in the distribution of dipolar bias and/or by an increase of the density of available vibrations with respect to an ordered solid. These results are not surprising, since previous studies on Mn_{12} molecules in frozen solutions³¹ and polymeric dispersions³² already showed a lowering of the magnetization relaxation but never justified the complete disappearance of the SMM behaviour. Finally, experiments performed on recrystallized sample B2 show also minor quantitative changes in the frequency-dependent response (see Fig. 6). Therefore, tentatively we discard acoustic phonon modifications or, in general, the lack of crystalline order, as the dominant physical cause responsible for the loss of SMM behaviour in our Mn_{12} deposits.

(II) Confinement effects upon nanostructuration

Surface effects on the magnetization relaxation of Mn_{12} SMMs upon surface deposition have already been studied.^{33,10} This is most likely our case due to the strong interaction of the first molecular layer with the surface and/or molecular strengths appearing at the outmost solid-air interphase. Such interactions prompted modulation of the molecular mechanical properties through structural modifications, as has recently shown by force-volume imaging experiments.¹⁹ This in turn will modify the axial anisotropy parameter (D) and their magnetization relaxation properties. Moreover, not only D but also other parameters such as the second-order transverse anisotropy terms (E) have been shown to be influenced. For instance, a recent study showed that the spin reversal time of standard Mn_{12}bz clusters decreases by 13% on reducing the crystal size from 11.5 to 0.4 μm (while maintaining constantly the same crystalline phase). The observed decrease in the effective energy barrier was associated with the existence of a distribution of E parameters, centred at $E = 0$, which broadens as the crystal size decreases and the surface/volume ratio of the particles increases.³⁴ While such effects are neither dominant in a crystal³⁵ nor in thick films deposited by different lithographic techniques,³⁶ in the case of deposition of a few layers they turn out to be crucial.

Conclusions

AFM-based lithography has been shown to be a suitable technique for the controlled integration of SMMs on magnetic sensors. The archetypal Mn_{12}bz system has been selectively deposited on a μ -SQUID sensor allowing for the first time the direct measurement of the frequency-dependent dynamical ac susceptibility of a few SMM layers. The results evidence the loss of the characteristic SMM behaviour, putting into evidence the relevance of nanoscale surface constraint effects when dealing with a few layers thick molecular film. Any future practical application out of SMMs will require overcoming these limitations by the use of more robust SMMs,³⁷ but more importantly, by preventing any modification that results from their deposition on surfaces.

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Notes and references

- 1 R. Sessoli, D. Gatteschi, A. Caneschi and M. A. Novak, *Nature*, 1993, **365**, 141–143; R. Sessoli, H. L. Tsai, A. R. Schake, S. Wang, J. B. Vincent, K. Folting, D. Gatteschi, G. Christou and D. N. Hendrickson, *J. Am. Chem. Soc.*, 1993, **115**, 1804–1816.
- 2 *Magnetochemistry*, ed. R. D. L. Carlin, Springer-Verlag, Berlin, 1986; *Magneto-Structural Correlations in Exchange Coupled Systems*, ed. R. D. Willet, D. Gatteschi and O. Kahn, Reidel Publishing, Dordrecht, The Netherlands, 1983.
- 3 J. Gomez-Segura, J. Veciana and D. Ruiz-Molina, *Chem. Commun.*, 2007, 3699–3707.
- 4 *Molecular Nanomagnets*, ed. D. Gatteschi, R. Sessoli and J. Villain, Oxford University Press, Oxford, UK, 2006.
- 5 J. R. Friedman, M. P. Sarachik, J. Tejada and R. Ziolo, *Phys. Rev. Lett.*, 1996, **76**, 3830–3833; L. Thomas, F. Lioni, R. Ballou, D. Gatteschi, R. Sessoli and B. Barbara, *Nature*, 1996, **383**, 145–147; J. M. Hernández, X. X. Zhang, F. Luis, J. Bartolomé, J. Tejada and R. Ziolo, *Europhys. Lett.*, 1996, **35**, 301–306.
- 6 D. Gatteschi and R. Sessoli, *Angew. Chem., Int. Ed.*, 2003, **42**, 268–297.
- 7 N. Domingo, E. Bellido and D. Ruiz-Molina, *Chem. Soc. Rev.*, 2012, **41**, 258–302.
- 8 M. Mannini, F. Pineider, P. Saintavit, C. Danieli, E. Otero, C. Sciancalepore, A. M. talarisco, M.-A. Arrio, A. cornia, D. Gatteschi and R. sessoli, *Nat. Mater.*, 2009, **8**, 194–197.
- 9 A. Saywell, A. J. Britton, N. Taleb, M. del Carmen Giménez-López, N. R. Champness, P. H. Beton and J. N. O'Shea, *Nanotechnology*, 2011, **22**, 075704.
- 10 M. Mannini, Ph. Saintavit, R. Sessoli, Ch. Cartier dit Moulin, F. Pineider, M.-A. Arrio, A. Cornia and D. Gatteschi, *Chem.-Eur. J.*, 2008, **14**, 7530–7535.
- 11 U. del Pennino, V. De Renzi, R. Biagi, V. Corradini, L. Zoppi, A. Cornia, D. Gatteschi, F. Bondino, E. Magnano, M. Zangrando, M. Zacchigna, A. Lichtenstein and D. W. Boukhvalov, *Surf. Sci.*, 2006, **600**, 4185–4189.
- 12 F. Moro, R. Biagi, V. Corradini, M. Evangelisti, A. Gambardella, V. De Renzi, U. del Pennino, E. Coronado, A. Forment-Aliaga and F. M. Romero, *J. Phys. Chem. C*, 2012, **116**, 14936–14942.
- 13 S. Voss, M. Fonin, U. Rüdiger, M. Burgert, U. Groth and Y. S. Dedkov, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **75**, 045102; M. Burgert, S. Voss, S. Herr, M. Fonin, U. Groth and U. Rüdiger, *J. Am. Chem. Soc.*, 2007, **129**, 14362–14366.
- 14 S. Kahle, Z. Deng, N. Malinowski, Ch. Tonnoir, A. Forment-Aliaga, N. Thontasen, G. Rinke, D. Le, V. Turkowski,



- T. S. Rahman, S. Rauschenbach, M. Ternes and K. Kern, *Nano Lett.*, 2012, **12**, 518–521.
- 15 L. Menga, B.-F. Lin, J.-Y. Yang, Y. Sun, R.-F. Dou, L.-X. Mab, Ch. M. Xiong and J.-C. Nie, *Chem. Phys. Lett.*, 2012, **542**, 81–84.
- 16 H. Sun, W. Li, L. Wollenberg, B. Li, L. Wu, F. Li and L. Xu, *J. Phys. Chem. B*, 2009, **113**, 14674–14680.
- 17 B. Fleury, V. Huc, L. Catala, P. Jegou, L. Baraton, Ch. David, S. Palacin and T. Mallah, *CrystEngComm*, 2009, **11**, 2192–2197.
- 18 F. Moro, V. Corradini, M. Evangelisti, R. Biagi, V. De Renzi, U. del Pennino, J. C. Cezar, R. Inglis, C. J. Miliosef and E. K. Brechin, *Nanoscale*, 2010, **2**, 2698–2703.
- 19 G. Otero, E. Evangelio, C. Rogero, L. Vázquez, J. Gómez-Segura, J. A. Martín Gago and D. Ruiz-Molina, *Langmuir*, 2009, **25**, 10107–10115.
- 20 M. Mannini, *et al.*, *Adv. Mater.*, 2009, **21**, 167; N. Grumbach, A. Barla, L. Joly, B. Donnio, G. Rogez, E. Terazzi, J.-P. Kappler and J.-L. Gallani, *Eur. Phys. J. B*, 2010, **73**, 103–108.
- 21 M. J. Martínez-Pérez, E. Bellido, R. de Miguel, J. Sesé, A. Lostao, C. Gómez-Moreno, D. Drung, T. Schurig, D. Ruiz-Molina and F. Luis, *Appl. Phys. Lett.*, 2011, **99**, 032504; E. Bellido, I. Ojea-Jiménez, A. Ghirri, Ch. Alvino, A. Candini, V. Puentes, M. Affronte, N. Domingo and D. Ruiz-Molina, *Langmuir*, 2012, **28**, 12400–12409.
- 22 E. Bellido, R. de Miguel, D. Ruiz-Molina, A. Lostao and D. MasPOCH, *Adv. Mater.*, 2010, **22**, 352–355; E. Bellido, R. d. Miguel, J. Sesé, D. Ruiz-Molina, A. Lostao and D. MasPOCH, *Scanning*, 2010, **32**, 35–41.
- 23 K. Takeda, K. Awaga, T. Inabe, A. Yamaguchi, H. Ishimoto, T. Tomita, H. Mitamura, T. Goto, N. Mori and H. Nojiri, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2002, **65**, 094424.
- 24 S. Bechstein, A. Kirste, D. Drung, M. Regin, O. Kazakova, J. Gallop, L. Hao, D. Cox and T. Schurig, *IEEE Trans. Appl. Supercond.*, 2013, **23**, 1602004.
- 25 M. J. Martínez-Pérez, J. Sesé, F. Luis, D. Drung and T. Schurig, *Rev. Sci. Instrum.*, 2010, **81**, 016108.
- 26 M. J. Martínez-Pérez, J. Sesé, F. Luis, R. Córdoba, D. Drung, T. Schurig, E. Bellido, R. de Miguel, C. Gómez-Moreno, A. Lostao and D. Ruiz-Molina, *IEEE Trans. Appl. Supercond.*, 2011, **21**, 345.
- 27 R. V. Martínez, F. García, R. García, E. Coronado, A. Forment-Aliaga, F. M. Romero and S. Tatay, *Adv. Mater.*, 2007, **19**, 291–295; B. Fleury, L. Catala, V. Huc, Ch. David, W. Z. Zhong, P. Jegou, L. Baraton, S. Palacin, P.-A. Albouyd and T. Mallah, *Chem. Commun.*, 2005, 2020–2022.
- 28 Z. Salman, K. H. Chow, R. I. Miller, A. Morello, T. J. Parolin, M. D. Hossain, T. A. Keeler, C. D. P. Levy, W. A. MacFarlane, G. D. Morris, H. Saadaoui, D. Wang, R. Sessoli, G. G. Condorelli and R. F. Kiefl, *Nano Lett.*, 2007, **7**, 1551–1555.
- 29 A. R. Farrell, J. A. Coome, M. R. Probert, A. E. Goeta, J. A. K. Howard, M.-H. Lemée-Cailleau, S. Parsons and M. Murrie, *CrystEngComm*, 2013, **15**, 3423–3429.
- 30 C. Carbonera, F. Luis, J. Campo, J. Sánchez-Marcos, A. Camón, J. Chaboy, D. Ruiz-Molina, I. Imaz, J. van Slageren, S. Dengler and M. González, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **81**, 014427.
- 31 A. Caneschi, T. Ohm, C. Paulsen, C. Sangregorio and R. Sessoli, *J. Magn. Magn. Mater.*, 1998, **177–181**, 1330; H. J. Eppley, H. L. Tsai, N. Devries, K. Folting, G. Christou and D. N. Hendrickson, *J. Am. Chem. Soc.*, 1995, **117**, 301.
- 32 F. El Hallak, J. van Slageren, J. Gómez-Segura, D. Ruiz-Molina and M. Dressel, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **75**, 104403.
- 33 L. Bogani, L. Cavigli, M. Gurioli, R. L. Novak, M. Mannini, A. Caneschi, F. Pineider, R. Sessoli, M. Clemente-León, E. Coronado, A. Cornia and D. Gatteschi, *Adv. Mater.*, 2007, **19**, 3906–3911.
- 34 N. Domingo, F. Luis, M. Nakano, M. Muntó, J. Gómez, J. Chaboy, N. Ventosa, J. Campo, J. Veciana and D. Ruiz-Molina, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **79**, 214404; M. Munto, J. Gómez-Segura, J. Campo, M. Nakano, N. Ventosa, D. Ruiz-Molina and J. Veciana, *J. Mater. Chem.*, 2006, **16**, 2612–2617.
- 35 L. Chen, W. Wernsdorfer, Ch. Lampropoulos, G. Christou and I. Chiorescu, *Nanotechnology*, 2010, **21**, 405504.
- 36 K. Kim, D. M. Seo, J. Means, V. Meenakshi, W. Teizer, H. Zhao and K. R. Dunbar, *Appl. Phys. Lett.*, 2004, **85**, 3872–3874; J. Means, V. Meenakshi, R. V. A. Srivastava, W. Teizer, Al. A. Kolomenskii, H. A. Schuessler, H. Zhao and K. R. Dunbar, *J. Magn. Magn. Mater.*, 2004, **284**, 215–219; K. Kim, A. Ford, V. Meenakshi, W. Teizer, H. Zhao and K. R. Dunbar, *J. Appl. Phys.*, 2007, **102**, 094306.
- 37 M. Mannini, F. Pineider, C. Danieli, F. Totti, L. Sorace, Ph. Saintavit, M.-A. Arrio, E. Otero, L. Joly, J. C. Cezar, A. Cornia and R. Sessoli, *Nature*, 2010, **468**(8), 417–421.

